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Unveiling the Mechanisms Behind Temperature-Dependent Impedance Behavior in PEM Fuel Cells at High Current Loads

PROBLEM

The necessity to move away from fossil fuels due to their role in global warming is an imperative. PEM Fuel Cells offer a clean energy alternative, but their widespread adoption is limited. A key challenge lies in understanding their microscale behavior, particularly the transport of gases and charge within the Membrane Electrode Assembly (MEA).

OBJECTIVE

This study aims to identify impedance behavior using electrochemical impedance spectroscopy across a wide range of temperatures and high current densities. Therefore, unveiling the mechanisms behind the cell's impedance, particularly at high current densities where concentration losses are significant.

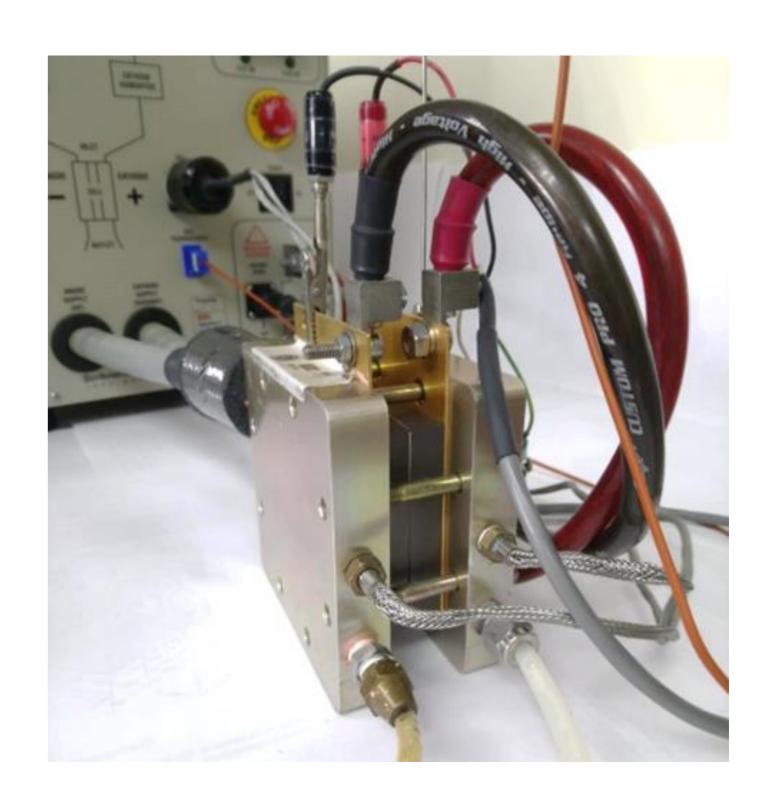


Fig. 1. Experimental PEM Fuel Cell

METHODS

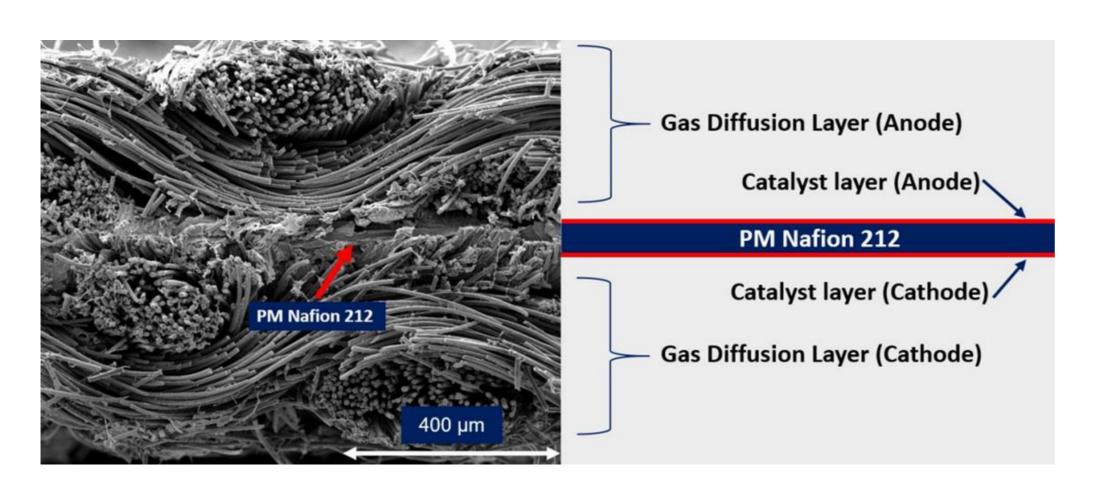


Fig. 2. Microscopic view of the Membrane

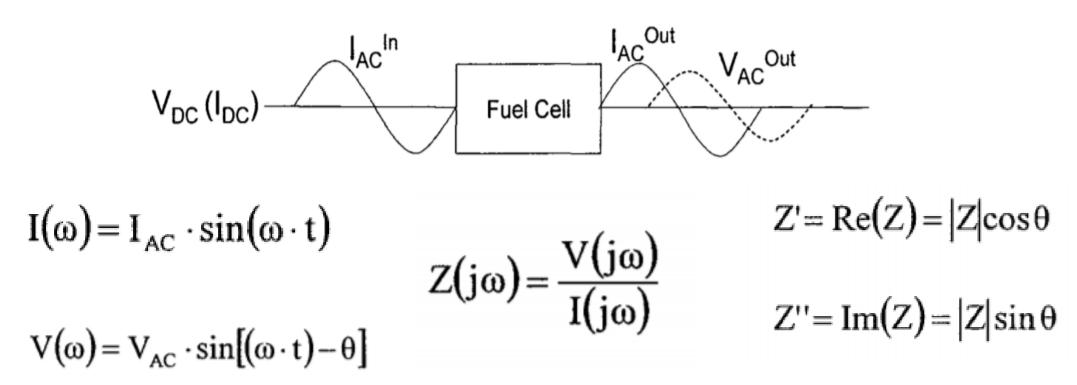
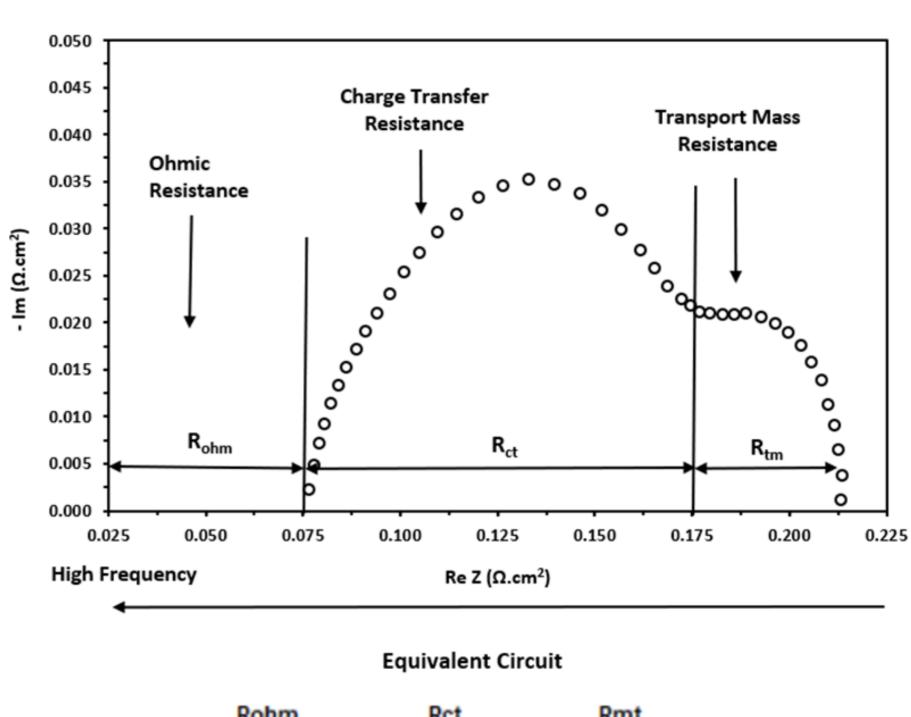


Fig. 3. Electrochemical Impedance Spectroscopy equations



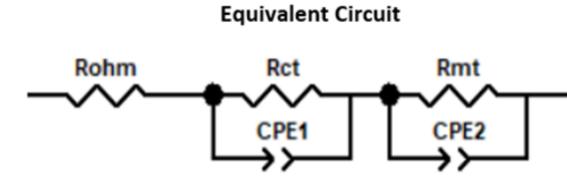


Fig. 4. Interpretation of Nyquist Diagram

RESULTS

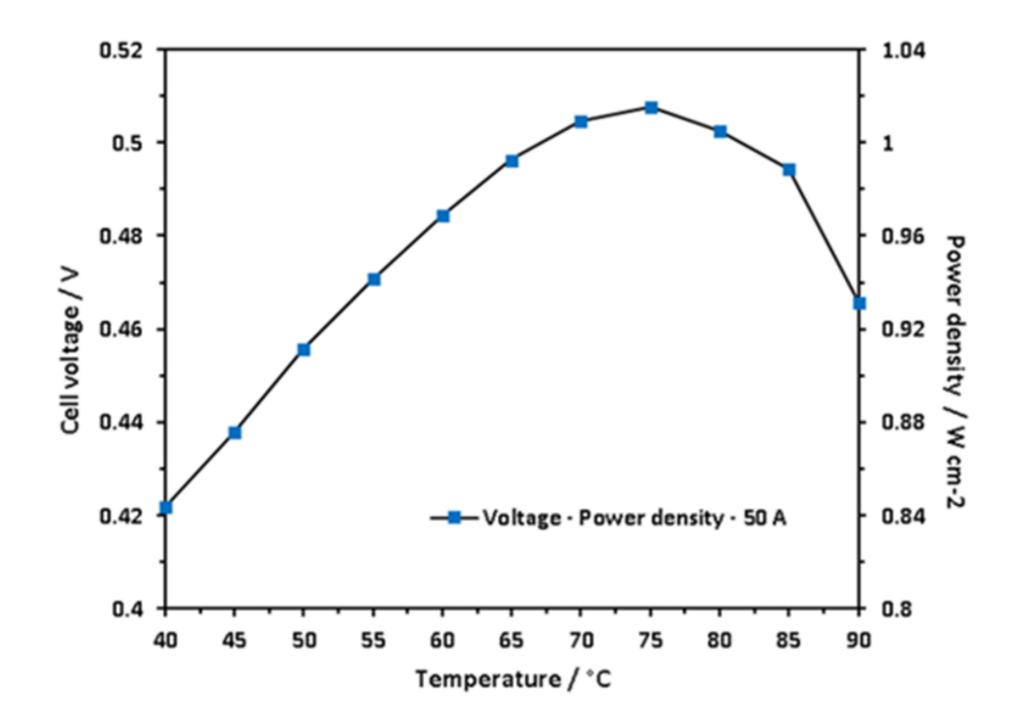


Fig. 5. Voltage-Power performance curve

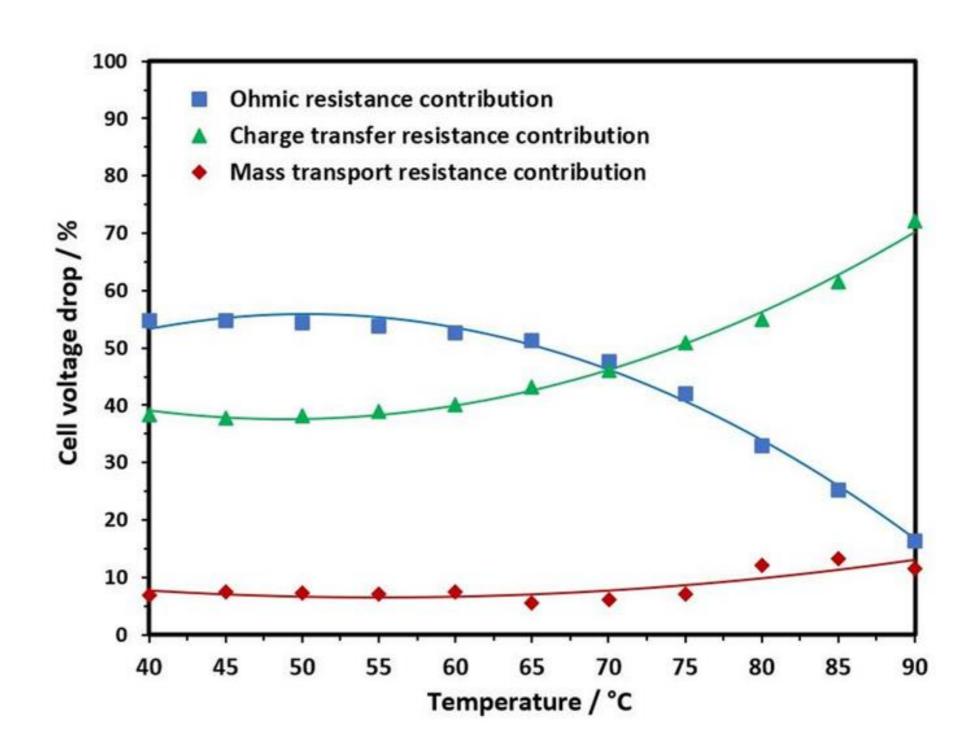


Fig. 6. Unveiling the individual polarization losses

CONCLUSIONS

- Ohmic Resistance is Dominant below 70°C, max 54 % at 40°C. Dependent on temperature and membrane hydration.
- Charge and Mass Transport Resistances have a minimal effect below 60°C. Charge transfer resistance significant above 60°C, max 72% at 90°C.
- Reducing charge transfer resistance crucial for works at high current densities.

ACKNOWLEDGMENTS

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